8	
9.	
10	
11.	
12. 13.	
13.	
Name Kly	Class
U	What is a Mineral?
Solid	

: s	· Cannot be a liquid or a gas
. †	NeturallyOccurring
N	• Found in <u>Nature</u> , not man-made
	Inorganic Is not <u>alive</u> and never was, <u>non-living</u>
1	• Is not <u>alive</u> and never was, <u>non-living</u>
	Formula
	Has a chemical <u>formula</u> , most are formed from
F	Companys of two or more elements, some
	minerals consist of one <u>eliment</u> ex. Au (Gold)
	Crystal Structure
С	• A definite <u>Structure</u> in which <u>atoms</u> are arranged

Date _____

Make your guesses in the first 2 columns, correct answers in the last two columns:

Mineral	Non-Mineral	Mineral	Non-Minera
	9		

Definition: A mineral is a naturally formed, inorganic solid that has a definite crystalline structure.

Name	Class	Date				
What is a Mineral?						

1. Two Groups of Minerals

- a. Silicates
 - i. Contains Silicon (Si) and Oxygen (O) 90% of Earth's crust
 - iii. Combine with elements such as Al, Fe, Mg, and K
- b. Non-silicates
 - i. Do not contain combinations of Si and O
 - Classes of Non-silicates
 - a. Native Elements: Au, Cu, Ag
 - b. Oxides: Al or Fe combine with O
 - c. Carbonates: C + O
 - d. Sulfates: S + O
 - . Odliates. G . G
 - e. Halides: F, CI, I, or Br combine with Na, K, or Ca
 - f. Sulfides: Pb. Fe. Ni, combine with S

2. Identifying Minerals

- a. Color the same mineral can come in a variety of colors, not reliable
- b. Luster the way in which a mineral reflects light
 - i. Metallic opaque and reflective, like metal
 - ii. Submetallic to nearly opaque and reflect well
 - iii. Nonmetallic not looking like a metal at all
 - 1. Vitreous glassy
 - 2. Silky fibrous
 - 3. Resinous yellow, dark orange, brown
 - 4. Waxv
 - 5. Pearly like inside a mollusk
 - 6. Earthy dull, looks like dirt or clay
- c. Streak the color of the mineral in a powdered form
- d. Cleavage when minerals break along a smooth, flat surface
- e. Fracture when minerals break unevenly along a curved or irregular surface
- f. Hardness a mineral's resistance to being scratched
 - Moh's Scale of Hardness 1 10
- G. Density how much matter is in a given amount of space (mass/volume)

H. Special Properties

- i. Fluorescence glow under ultraviolet light
- ii. Chemical Reaction fizzes with acid test
- iii. Optical Properties double image formed
- iv. Magnetism magnetic, attract iron
- v. Taste saltv
- vi. Radioactivity radium or uranium contained